

## Triple Chemistry:Year 10

| Scheme of Learning   | Assessments  |
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| <ul> <li>AO1: Demonstrate knowledge and understanding<br/>of: scientific ideas; scientific techniques and<br/>procedures.</li> </ul>   | All Tests and Required Practicals cover all the assessment objectives. |
| <ul> <li>AO2: Apply knowledge and understanding of:<br/>scientific ideas; scientific enquiry, techniques<br/>and procedures.</li> </ul>  |  |
| <ul> <li>AO3: Analyse information and ideas to: interpret<br/>and evaluate; make judgments and draw<br/>conclusions; develop and improve experimental<br/>procedures.</li> </ul> |  |
| Half Term One:   | Assessment 1   |
| The Overarching Inquiry:   | Test C4.1 Atomic Structure   |
| What are atoms?  | Assessment 2   |
| C4.1 Atomic structure  | Literacy Assessment- Mendeleev<br>Periodic Table                       |
| C4.1 Atomic structure  | renouic rable  |
| The Overarching Inquiry:<br>What is a chemical reaction and how can we<br>calculate its purity?  |  |
| C4.3 Quantitative Chemistry  |  |
| Half Term Two:   | Assessment 3   |
|  | Test C4.3 Quantitative Chemistry                                       |
| The Overarching Inquiry:<br>What is a chemical reaction and how can we   | Assessment 4   |
| calculate its purity?  | RP Soluble salt  |
| C4.3 Quantitative Chemistry  | Assessment 5<br>RP Titrations  |

| The Overarching Inquiry:<br>How can the physical conditions effect the rate<br>of a chemical reaction?<br>C4.4 Chemical Changes   |  |
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| <ul> <li>Half Term Three:</li> <li>The Overarching Inquiry:</li> <li>How can we use our knowledge of chemical reactions to make new materials?</li> <li>C4.4 Chemical Changes</li> <li>The Overarching Inquiry:</li> <li>How can chemical reactions be useful in everyday life?</li> <li>C4.5 Energy Changes</li> </ul> | Assessment 6<br>RP Electrolysis<br>Assessment 7<br>C4.4 Chemical Changes<br>Assessment 8<br>RP Investigating temperature changes<br>Assessment 9<br>C4.5 Energy Changes  |
| Half Term Four:<br>The Overarching Inquiry:<br>How can the physical conditions effect the rate<br>of a chemical reaction?<br>C4.6 The rate and extent of chemical change  | Assessment 10<br>RP Investigating the effect of<br>temperature mon the rate of a<br>chemical reaction.<br>Assessment 11<br>RP Investigating the effect of<br>concentration on the rate of a<br>chemical reaction.<br>Test C4.6 The rate and extent of<br>chemical change |
| Half Term Five:<br>The Overarching Inquiry:<br>How can organic molecules be used to make new<br>and useful materials?<br>C4.7 Organic Chemistry   | Assessment 11<br>C4.7 Organic Chemistry<br>Assessment 12<br>RP Chromatography  |

| Half Term Six:                          | Assessment 13               |
|---|-----------------------------|
|   | RP Identification of Ions   |
| The Overarching Inquiry:                |                             |
| How can we identify specific chemicals? | Assessment 14               |
|   | Test C4.8 Chemical Analysis |
| C4.8 Chemical Analysis                  |                             |
|   | End of Year Exams           |
| Revision and DTT                        |                             |
|   |                             |